

NOTES

**New Cytotoxic Bafilomycin C₁-amide
Produced by *Kitasatospora cheerisanensis***

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The genus *Kitasatospora* comprises a group of filamentous and aerobic Gram-positive bacteria and is phenotypically closely related to *Stereptomyces* strains. Characteristics of the genus *Kitasatospora* are the presence of *meso*-diaminopimelic acid, *LL*-diaminopimelic acid, and galactose in the cell wall.^{1,2)} During the course of our search for antifungal metabolites from microbes, we found that methanolic extracts of the culture broth of a new species of the genus *Kitasatospora* isolated from a soil sample inhibited the growth of plant pathogens, such as *Rhizoctonia solani*, *Botrytis cinerea*, and *Phythium ultimum*. Classification and characteristics of the species have been reported.³⁾ In this paper, we report the isolation, structure determination, and biological activities of a new bafilomycin derivative **2** designated bafilomycin C₁-amide as well as the known bafilomycins, **1**, **3**, **4**, and **5** from the fermentation broth.

ExperimentalMicroorganism

The producing microorganism was a newly identified isolate as a strain of *Kitasatospora cheerisanensis* YC75

collected from the Cheerisan mountain located at the southern part of Korea (deposited at the Korean Collection for Type Cultures as KCTC 2395).³⁾

Fermentation

A stock culture of *K. cheerisanensis* YC75 was cultured on potato dextrose agar at 28°C for 7 days and two or three loops of the bacteria from the agar was inoculated to a 1-liter Erlenmeyer flask containing tryptic soy broth (200 ml) as a seed medium. The flask was then incubated at 28°C for 3 days on a rotary shaker (180 rpm) and the whole culture broth was transferred into a 5-liter fermenter containing production medium (3 liters) composed of glucose 0.5 g, soluble starch 20 g, beef extract 1 g, soybean flour 25 g, K₂HPO₄ 0.5 g, NaCl 2 g, CaCO₃ 6 g, and yeast extract 2 g in 1 liter of distilled water. The fermentation was carried out at 28°C for 4 days with agitation at 500 rpm and aeration at 3 liters/minute.

Biological Assays

The paper disc diffusion assay was used for measuring the antifungal activity and the protein-binding sulforhodamine B (SRB) method for the cytotoxicity against cancer cell lines.^{4,5)}

Analytical Procedures

The UV absorption spectra were obtained on a Waters 996 Photodiode Array Detector during HPLC. The IR spectra were taken in KBr pellets using a Perkin Elmer BX FT-IR spectrometer. The mass spectra were obtained on a JEOL HX110A-HX110A Tandem HR mass spectrometer. The NMR spectra for ¹H and ¹³C were obtained on a Varian Unity-500 NMR spectrometer at 500 MHz and 125 MHz, respectively. The chemical shifts given in ppm were referenced to solvent peaks: δ_{H} 3.30 and δ_{C} 49.0 for methanol-*d*₄; δ_{H} 2.05 and δ_{C} 29.9 (methyl) for acetone-*d*₆.

Isolation

The fermentation broth (5 liters, pH 7.2) was centrifuged at 4,000 *g* and cell cake was collected (424 g, wet). The cell cake was extracted with methanol (1 liter×3). The methanolic solutions combined were concentrated to

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dryness, which was suspended in methanol (500 ml) and filtered. The filtrate (13 g) was concentrated and suspended in distilled water and chromatographed (Merck, C-18, 43~60 μm , 30 \times 300 mm) under medium-pressure with a continuous gradient elution of water to methanol (10 ml/minute) to obtain 18 fractions (about 200 ml/each fraction). The 15th fraction (methanol-eluate) active against *Rhizoctonia solani* was concentrated to give an orange solid (669 mg), which was applied onto a column of Toyopearl HW-40S (15 \times 950 mm). The column was eluted with methanol to obtain 80 tubes (4 ml/each) and each tube was assayed against *R. solani*. Two groups (12th to 22nd and 23rd to 32nd) of the active fractions pooled were concentrated to obtain orange solids, 223 mg and 80 mg, respectively. Final purification of the active compounds was achieved by reverse-phase HPLC (20 \times 250 mm, Pegasil ODS, Shensu PAK). From the above 223 mg-sample, using continuous gradient of 60% aqueous acetonitrile to 100% acetonitrile for 130 minutes with flow of 7 ml/minute, compound **5** (3.5 mg), **4** (7.0 mg), **2** (4.7 mg), and **3** (20 mg) were eluted in the retention time of 55.5, 65.9, 69.3, and 107.9 minutes, respectively. The above 80-mg sample afforded compound **1** (17 mg) as an orange solid using continuous gradient of 4:1:5 to 2:2:1 acetonitrile/methanol/water for 40 minutes in the retention time of 17.0 minutes.

The compound **2** showed the following physicochemical properties: yellow oil; UV λ_{max} 244.3, 283.3 nm; IR (KBr) 3420, 2923, 2867, 1720, 1684, 1453, 1381, 1249, 1169, 1101, 917 cm^{-1} ; (+) HRFAB-MS m/z 742.4167 $[\text{M}+\text{Na}]^+$ (calcd for $\text{C}_{39}\text{H}_{61}\text{NO}_{11}\text{Na}$, 742.4142), 643.0, 537.1, 413.0, 175.9; ^1H and ^{13}C NMR data (acetone- d_6) are listed in Table 1.

Results

Isolation

The culture filtrate was extracted with methanol. The filtrate was fractionated on C-18 flash column chromatography. The bioactive fractions were subjected to a series of chromatographic techniques such as size exclusion and prep C-18 HPLC to afford compounds **1**~**5**. Analytical HPLC of the compounds **1**, **5**, **4**, **2**, and **3** showed the retention times of 5.7, 22.8, 28.6, 29.8, and 52.6 minutes, respectively, on a C-18 column (4.6 \times 250 mm, Pegasil ODS, 0.7 ml/minute, 70% aqueous acetonitrile to 100% acetonitrile for 60 minutes).

Structure-Elucidation

The molecular formula of compound **1** was deduced to be $\text{C}_{39}\text{H}_{60}\text{O}_{12}$ as $[\text{M}+\text{Na}]^+$ was observed at m/z 743.4022 (calcd for $\text{C}_{39}\text{H}_{60}\text{O}_{12}\text{Na}$, 743.3982) in the positive high resolution FAB mass spectrum. The UV absorption maxima at 245.5 and 284.5 nm and the IR absorptions at 1720, 1686, 1586, 1100 cm^{-1} suggest the presence of a conjugated diene and conjugated diene ester moiety. The ^{13}C NMR spectrum (methanol- d_6) of compound **1** contained thirteen sp^2 -carbons including three C=O groups at δ 172.6, 168.1, 167.6, 145.6, 145.5, 143.3, 142.3, 134.8, 134.6, 133.2, 128.8, 126.9, 125.5 ppm. In addition, nine oxygen-bearing sp^3 -carbons with a hemiketal functionality were observed at δ 100.3, 84.4, 81.2, 77.7, 77.3, 75.5, 72.0, 60.6, 55.9 ppm. The DEPT experiment showed the presence of eleven methyl groups at 60.6, 55.9, 22.5, 21.8, 20.1, 17.9, 14.7, 14.1, 12.6, 10.5, 7.25 ppm. The former two resonances were readily interpreted as methoxy groups. Combination of extensive 2-D NMR experiments (^1H -COSY, TOCSY, ROSEY, HSQC, and HMBC) provided the total structure of compound **1** which was previously reported as bafilomycin C_1 .^{6,7)}

The molecular formula of compound **2** was determined to be $\text{C}_{39}\text{H}_{61}\text{NO}_{11}$ as $[\text{M}+\text{Na}]^+$ was observed at m/z 742.4167 (calcd for $\text{C}_{39}\text{H}_{61}\text{NO}_{11}\text{Na}$, 742.4142) in the positive high resolution FAB mass spectrum. The UV absorption and IR spectra exhibited the characteristic diene and diene ester functionalities as in compound **1**. In addition, Compound **2** showed a quite similar pattern to compound **1** in the ^1H and ^{13}C NMR spectra except the appearance of exchangeable protons at δ 4.05, 4.77, 5.42, 6.74, and 7.32 ppm in the ^1H NMR spectrum in acetone- d_6 of **2**, as shown in Table 1. In particular, two broad signals at 6.74 and 7.32 ppm indicated the presence of amide functionality and comparison of the NMR data of the fumarate moieties suggested assignments of the amide functionality (Figure 1). Thus, the structure of compound **2** was deduced to be an amide form of compound **1** shown in Figure 2 and fully supported by extensive interpretation of 2-D NMR experiments with combination of high resolution FAB mass spectral data. We suggest compound **2** designated as bafilomycin C_1 -amide.

The molecular formula of compound **3** was determined to be $\text{C}_{44}\text{H}_{65}\text{NO}_{13}$ as $[\text{M}+\text{Na}]^+$ was observed at m/z 838.4370 (calcd for $\text{C}_{44}\text{H}_{65}\text{NO}_{13}\text{Na}$, 838.4354) in the positive high resolution FAB mass spectrum. The UV absorption maxima and IR absorptions were almost the same as those of compounds **1** and **2**, except a strong IR absorption band at 1620 cm^{-1} . The exchangeable proton

Table 1. ^1H and ^{13}C NMR chemical shifts of compounds **1** and **2**^a.

Position	1		2	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}
1		168.1		167.7
2		142.3		141.8
2-OCH ₃	3.61 (s)	60.6	3.62 (s)	60.1
3	6.70 (s)	134.8	6.70 (br s)	134.1
4		133.2		132.7
4-CH ₃	1.97 (s)	14.1	1.97 (d, 1.0)	14.1
5	5.88 (d, 9.3)	145.6	5.95 (d, 9.3)	145.6
6	2.52 (m)	38.5	2.55 (m)	38.0
6-CH ₃	1.05 (d, 6.8)	17.9	1.04 (d, 6.8)	17.7
7	3.22 (m)	81.2	3.30 (m)	80.3
7-OH			4.05 (d, 4.9)	
8	1.84 (m)	41.7	1.88 (m)	41.7
8-CH ₃	0.92 (d, 6.8)	22.5	0.91 (d, 6.8)	22.2
9	2.01 (m)	42.5	2.05 (m)	42.2
10		145.5		144.7
10-CH ₃	1.87 (s)	20.1	1.92 (s)	20.3
11	5.76 (d, 10.8)	125.5	5.79 (d, 10.8)	125.1
12	6.59 (dd, 14.7, 10.8)	134.6	6.67 (dd, 14.7, 10.8)	134.3
13	5.10 (dd, 14.7, 8.8)	126.9	5.13 (dd, 14.7, 9.3)	126.9
14	3.98 (dd, 8.8, 8.3)	84.4	4.05 (dd, 9.3, 8.3)	83.3
14-OCH ₃	3.24 (s)	55.9	3.22 (s)	55.6
15	5.02 (d, 8.30)	77.7	4.95 (dd, 8.3, 1.5)	77.3
16	2.07 (m)	39.4 ^b	2.16 (m)	38.2
16-CH ₃	0.86 (d, 6.8)	10.5	0.87 (d, 6.8)	10.3
17	4.15 (dd, 10.7, 1.0)	72.0	4.17 (m)	71.5
17-OH			4.77 (d, 4.0)	
18	1.80 (m)	43.7	1.85 (m)	42.9
18-CH ₃	0.98 (d, 7.3)	7.3	0.99 (d, 6.8)	7.4
19		100.3		99.7
19-OH			5.42 (d, 2.0)	
20	2.29 (dd, 11.7, 4.9)	40.5	2.33 (dd, 11.7, 4.9)	40.6
	1.31 (t, 11.7)		1.29 (td, 11.7, 2.0)	
21	4.99 (td, 11.7, 4.9)	75.5	5.02 (td, 10.8, 4.9)	75.7
22	1.63 (m)	39.4 ^b	1.61 (m)	39.0
22-CH ₃	0.83 (d, 6.4)	12.6	0.83 (d, 6.8)	12.5
23	3.58 (dd, 10.3, 2.0)	77.3	3.61 (dd, 10.7, 2.0)	76.4
24	1.92 (m)	29.3	1.93 (m)	28.7
24-CH ₃	0.96 (d, 6.8)	21.8	0.95 (d, 6.8)	21.6
25	0.82 (d, 6.8)	14.7	0.81 (d, 6.8)	14.6
1'		167.6		165.3 ^c
2'	6.48 (d, 15.9)	128.8	6.67 (d, 15.5)	130.6
3'	6.91 (d, 15.9)	143.3	7.05 (d, 15.5)	137.8
4'		172.6		165.5 ^c
4'-NH ₂			7.32 (br s)	
			6.74 (br s)	

^a Measured in methanol-*d*₄ for **1** and in acetone-*d*₆ for **2** at 24°C at 500 MHz (^1H) and at 125 MHz (^{13}C).

^b Overlapped.

^c Interchangeable assignments.

Fig. 1. Chemical shift assignments (δ_{H} , 6.48~7.58 ppm in italics; δ_{C} , 128.8~172.6 ppm) for the fumarate moiety A, B, and C of compounds **1**, **2**, and **3**, respectively.

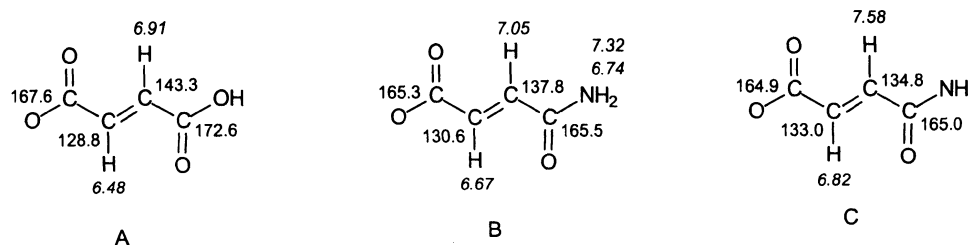
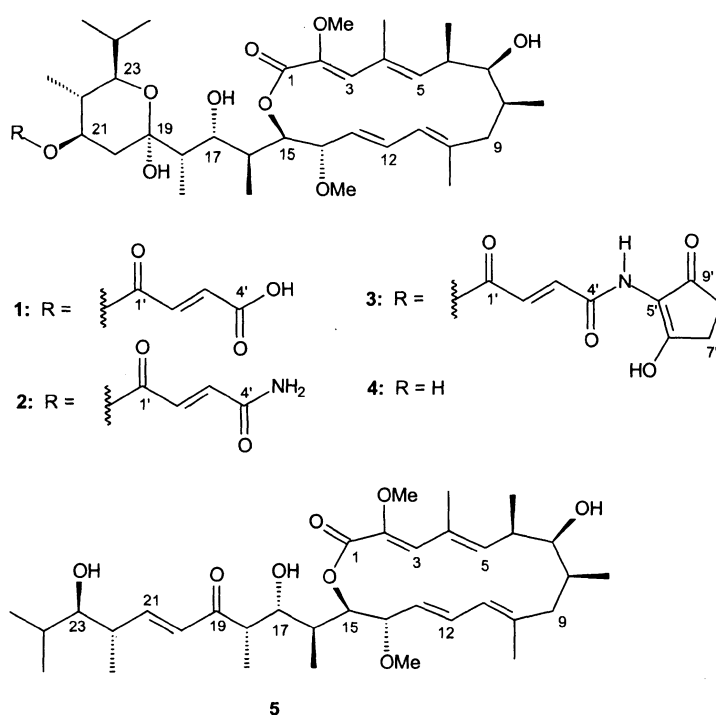


Fig. 2. Structures of compounds **1**~**5** isolated from *Kitasatospora cheerisanensis*.Table 2. Cytotoxicity (IC₅₀, ng/ml) of compounds against human tumor cell lines.

Cell line ^a	Compound				
	1	2	3	4	Adriamycin
A-549	0.03	1.08	0.18	1.86	170
SK-OV-3	0.02	0.91	0.09	1.64	530
SK-Mel-2	0.008	0.20	0.02	1.93	260
XF-498	0.007	0.28	0.03	0.81	840
HCT-15	0.02	0.32	0.02	1.70	960

^a Cell lines: A-549, non-small cell lung cancer; SK-OV-3, ovarian cancer; SK-Mel-2, melanoma; XF-498, central nerve system cancer; and HCT-15, colon cancer.

signals at 6.74 and 7.32 ppm of compound **2** disappeared in compound **3**. The methine signals in the fumarate moiety showed remarkable difference in both ¹H and ¹³C NMR spectra, whose chemical shifts are depicted in Figure 1. In addition, newly found was the signals for 3-hydroxy-2-cyclopentenone-2-yl group. Thus, the structure of compound **3** was determined to be bafilomycin B₁.^{6,7)}

The molecular formula of compound **4** and **5** was determined to be C₃₅H₅₈O₉ and C₃₅H₅₆O₈, respectively, as [M+Na]⁺ was observed at *m/z* 645.3986 (calcd for C₃₅H₅₈O₉Na, 645.3979) and *m/z* 627.3871 (calcd for C₃₅H₅₆O₈Na, 627.3873) in the positive high resolution FAB

mass data, respectively. UV and IR absorption spectrum confirmed the presence of the same 16-membered macrolide lactone as in compounds **1**~**3**. The structures of compound **4** and **5** were determined to be bafilomycin A₁ and bafilomycin D, respectively, from the interpretation of extensive 2-D NMR experiments with combination of high resolution FAB mass spectral data and by comparison of the ¹H and ¹³C NMR assignments with literature values.^{6~8)}

Biological Activity

Cytotoxicity was measured against several human tumor

Table 3. Antifungal activity (inhibition zone, mm) of compounds against various plant-pathogens^a.

Test fungus	Compound			
	1	2 ^b	3	4
<i>Rhizoctonia soliani</i>	12	NT	10	2
<i>Botrytis cinerea</i>	15	NT	6	8
<i>Sclerotinia sclerotiorum</i>	6	NT	5	6
<i>Botryosphaeria dothidea</i>	10	NT	5	0
<i>Collectotrichum cingulata</i>	10	NT	10	0
<i>Magnaporthe griesa</i>	8	NT	6	2
<i>Phythium ultimum</i>	10	NT	6	5

^a Each compound (250 µg) was loaded onto each paper disc, which was placed on a PDA medium of test fungi. After incubation for 5 days, the inhibitory zone outside the paper disc was measured in mm.

^b NT: not tested because of insufficient material.

cellular models using SRB method.^{4,5} All the compounds tested were extremely cytotoxic against non-small cell lung cancer; ovarian cancer; melanoma; central nervous system cancer, and colon cancer cell lines in the range of 0.008 to 1.93 ng/ml (Table 2). Compound **1** (bafilomycin C₁) was the most potent in cytotoxicity. Compound **2** (bafilomycin C₁-amide) was cytotoxic at the concentrations of 0.20~1.08 ng/ml, which are over 100-fold more potent than adriamycin. The presence of the fumarate moiety at C21 seems to attribute to higher cytotoxicity.

The antifungal activity of compounds **1**, **3** and **4** was measured against plant pathogens (Table 3). Compound **1** showed the strongest activity. As in the case of cytotoxicity, the fumarate functionality seems to attribute to higher antifungal activity.

Discussion

The bafilomycins, a group of 16-membered macrolide lactones, have been isolated from several *Sterptomyces* species^{8~14}) and *Kitasatospora setae*.^{15,16}) This macrolide lactone moiety has been found in some other metabolites such as hygrolidins,¹⁷) L-155,175,¹⁸) leucanicidins,¹⁹) PD 118,576s,^{20,21}) and micromonosporides.^{22,23}) These macrolides show a variety of biological activities such as antifungal, antiparasitic, and antitumor activity as well as inhibitory activity against Na⁺, K⁺ activated ATPase²⁴) or gastrulation of the starfish embryo. Bafilomycin A₁ inhibits cell growth through apoptosis.^{25~28}) Bafilomycin C₁ is the most potent in antibacterial activity⁹) and bafilomycin B₁ in nematode development inhibitory activity among natural

bafilomycins.¹²) The absolute stereochemistry of bafilomycin A₁ has been established by the X-ray crystallographic analysis and NMR spectroscopy.^{29,30}) Those remarkable biological properties have stimulated great interest in synthesis.^{31,32}) Structure-activity relationship of bafilomycin A₁ analogues shows that the parent bafilomycin A₁ is most potent in the vacuolar ATPases than any of the new analogues.³³)

Bafilomycins (**1~5**) were purified from the fermentation of a new isolate *Kitasatospora cheerisanensis*. Compound **2** (bafilomycin C₁-amide) has not been yet reported, although it is not clear whether **2** is derived from **3** (bafilomycin B₁) during isolation. Compounds (**1~4**) tested showed potent cytotoxicity against human cancer cell lines, which was significantly dependent on the substitutional change at 21-OH. Compound **1** (bafilomycin C₁) was the most potent. The fumarate ester functionality at C21 leads to more significant cytotoxicity (**1**, **2**, and **3** vs **4**) while change of functionality from carboxylic acid to amide at the fumarate moiety resulted in decrease in cytotoxicity as shown between **1** and **2**. Compound **1** was the most potent against plant pathogens tested. It is noteworthy that the fumarate ester moiety in micromonosporides does not affect the activity of gastrulation of starfish embryos.²³) Structural modification at C21-OH would give valuable structure-activity information.

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